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OIL and GAS DIVISION

1 6 JUN 1983

## PETROLEUM GEOCHEMISTRY



HYDROCARBON SOURCE ROCK

CHARACTERIZATION STUDY

VOLADOR No. 1 WELL

Prepared for

SHELL DEVELOPMENT (AUSTRALIA) PTY. LTD.

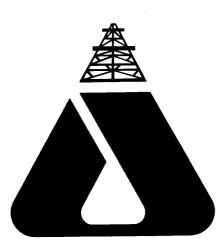
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52 MURRAY ROAD, WELSHPOOL, W.A. 6106. Telephone (09) 458 7999 Telex: ANALAB AA92560

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### OIL and GAS DIVISION

### 1 6 JUN 1983

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### CHARACTERIZATION STUDY

VOLADOR No. 1 WELL

### SUMMARY

Organic geochemical analyses carried out on sidewall cores, crude oils, water and gas samples from the well interval 3549.9m to 4554m in the Volador No. 1 well have indicated the following:-

- Marginally mature rocks are encountered around 3700m to 3800m and mature rocks are penetrated between 3800m to 4100m. At bottom of the analyzed well interval, 4554m, the sediments are interpreted to still be oil-generative.
- The rocks in well interval 3549.9m to 4039.0m are extremely organic rich, but due to the low thermal maturity of the sediments, it is doubtful that significant petroleum generation and expellsion has occurred.
- The rocks in well interval 4145.3m to 4554m are marginally mature to mature good to excellent petroleum source sediments. In particular those sediments within interval 4526m to 4554m have very favourable oil source characteristics.
- The condensate and crude oils produced from this well are low sulphur, very waxy liquid hydrocarbons. We interpret that the condensate is a result of a natural separation of the crude oil within the reservoirs.
- The good oil source rocks encountered between 4526m to 4554m are similar to these oils and maybe the parent source.
  - The three crude oils analyzed from Halibut A-1, Hapuku 1 and Tuna A-2 wells appear to be genetically related, and are also similar to the oils produced from the Volador No. 1 well.

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PAUL TYBOR Manager - Operations

### INTRODUCTION

Organic geochemical analyses have been performed on sixteen (16) side wall core samples, seventeen (17) crude oil samples, three (3) gas cylinder samples and eighteen (18) water samples from the Volador No. 1 well, drilled in Gippsland Basin, offshore Australia. The crudes, waters and gas samples were recovered during two production tests run on the well. In addition to these samples, three oil samples from the Halibut Al, Hapuku 1 and Tuna A-2 wells were also analyzed for characterization.

This study was undertaken to evaluate the hydrocarbon source quality (oil vs gas) richness and state of thermal maturity (pre oil, oil generative, eometamorphosed) of the sediments penetrated between well interval 3549.9m to 4554m in the Volador No. 1 well. The oils, waters and gases were analyzed to characterize the fluids and gases produced during the testing of the well. The oils were compared to the three oil samples from the Hapuku 1, Halibut A-1 and Tuna A-2 wells to determine if any genetic relationships exists between the samples.

The results of the analyses carried out on these samples are presented in the following:-

Type of Analysis	Figure	Table
Rock Samples		
% Total organic carbon determination	1	1
Pyrolysis analysis	1	1
Extraction and liquid chromatography	2	2
C <sub>12</sub> + saturate gas chromatography	3; 4A to 4N	3
Vitrinite reflectance	1 - 3	4
Crude Oils		
API gravity, % sulphur, viscosity, pour point		5
C <sub>12</sub> + saturate gas chromatography (Sample 8)	5	6
Whole oil gas chromatography (Volador samples)	6A to 6D	7
Whole oil gas chromatography (Gippsland Oils)	7A to 7C	8

### Type of Analysis

Figure

### <u>Table</u>

9

10

Gas Samples

Complete gas analysis ( 3 cylinder samples)

Stable carbon isotopic determination

### Water Samples

Complete formation water analysis (18 water samples)

11A to 11R

Initially  $C_{12}$  + saturate gas chromatography was planned for evaluation of the crude oil samples. However, this approach was reconsidered, and it was decided to carry out whole oil gas chromatography on the samples for better characterisation. Consequently, only crude oil sample No. 8 was submitted to  $C_{12}$  + saturate gas chromatography, with whole gas chromatography analysis performed on all crude oil samples.

A description of the analyses performed on the rock and crude oil samples in this well is presented in Appendix I, located at the back of this report.

### General Information

Copies of this report have been mailed to Mr. Steve Rigby of Shell Development (Australia) Pty. Ltd. Any questions regarding this study can be directed to either Mr. Paul Tybor or Dr. Garry Woodhouse of Analabs in Perth Western Australia.

All data and interpretations given herein are proprietary to the Shell Development (Australia) Pty. Ltd., and are treated as highly confidential material by all Analabs personnel.

### **RESULTS AND INTERPRETATIONS**

### A. <u>Thermal Maturity of Sediments</u>

Based on vitrinite reflectance measurements the sediments from 3549.9m to 3820.3m are marginally mature and are in the early stages of petroleum generation. Below 3820.3m the recorded reflectances are considered mature values and these sediments are interpreted to be oil generative (Figures 1-3; Table 4).

The results for pyrolysis (Tmax and Production Index; Figure 1; Table 1) analysis indicate a slightly less mature geothermal history for this sedimentary sequence. The following breaks out the maturation zones of this interval, as determined by pyrolysis:

<u>Interval</u>		<u>a1</u>	Tmax Range	P.I. Range	Interpretation	
3549.9m	-	3691.5m	423° - 429°	0.03 to 0.07	Immature	
3799m	-	4039m	433° - 441°	0.04 to 0.08	Marginally mature	
4145.3m	-	4554.Om	443° - 451°	0.14 to 0.26	Mature	

The  $C_{12}^{+}$  saturate gas chromatography data appears to correspond more Closely to maturities determined by pyrolysis than by vitrinite reflectance. The carbon preference index values indicate immaturity (CPI (1) > 1.5) from 3549.9m to 3820.3m, marginally mature (CPI (1) 1.2 - 1.5) from 3820.3m to 4145.3m and mature (CPI (1) 1.0 - 1.2) sediments from 4145.3m to 4554.0m (Figure 3; Table 3). The chromatograms themselves exhibit immature characteristics with isoprenoid pristane predominating over the normal alkanes in samples down to 4360m (Figures 3; 4). This immature feature may also be a function of the type of organic matter contained in these sediments, which give immature  $C_{12}^{+}$  saturate characteristics at moderately mature to mature maturation levels. Since these rocks contain large amounts of terrestrial organic matter, more time and temperature may be required before oil generation occurs.

Regardless of which data is utilised in assessing the thermal maturity of the sediments analysed from this well, it appears that marginally mature rocks are encountered around 3700m to 3800m, while mature sediments are penetrated between 3800m to 4100m. At the bottom of the analyzed well interval the sediments are interpreted to still be oil generative.

### B. Hydrocarbon Source Characterisation of Sediments

### Well Interval 3549.9m to 4039.0m

Well interval 3549.9m to 4039.0m is comprised of very carbonaceous claystones, siltstones and coals that contain extremely rich concentrations of organic matter (% TOC; Figure 1; Table 1). Due to the immature nature of these

### B. Hydrocarbon Source Characterisation of Sediments (Cont)...

### Well Interval 3549.9m to 4039.0m

rocks and the apparent abundance of terrestrial organic matter types dispersed within these sediments, the majority of the high amounts of free hydrocarbon (S1; Figure 1; Table 1) and extracted hydrocarbon (Figure 2; Table 2) is predominantly aromatic in nature. Since crude oils are usually comprised of greater than 80% saturate hydrocarbon, and low amounts of aromatic hydrocarbon, it is doubtful that these sediments are capable of generating significant quantities of oil. Presently, these rocks are interpreted to be prospective for large volumes of indigenously generated gas. At more mature levels of thermal maturation, oil generation and expulsion could have occurred.

### Well Interval 4145.3m to 4554.0m

The rocks in well interval 4145.3m to 4554.0m contain good to excellent amounts of apparently mixed oil and gas prone organic matter, at moderately mature to mature maturation levels. Good to excellent amounts of extractable organic matter and  $C_{1,2}$ + total hydrocarbon (Figure 2; Table 2), and good to excellent amounts of free hydrocarbon  $(S_1;$  Figure 1; Table 1) were analysed from these sediments. This hydrocarbon contains greater amounts of saturate hydrocarbon than aromatic, and thus appears to have favourable oil generating capabilities. This favourable oil source character is very evident in samples at 4526m, 4536m and 4554m, where the  $C_{12}$ + saturate gas chromatogram configurations approach those of the whole oil gc's of the value of the saturate configuration of the saturate configurate configuration of the saturate configuratic configuration of t the oils produced from this well, and the other Gippsland oils included in this study (Figures 6 and 7). The oils recovered from this well were produced from intervals 3756m There are very similar to 3783m and 3911m to 3914m. characteristics between these oils and the source rocks encountered at 4524m 4536m and 4554m. These similarities will be compared and discussed in the following section entitled crude oil characterisation.

As a result, the rocks in well interval 4145.3m to 4554.0m are marginally mature to mature, good to excellent petroleum source sediments. In particular, those sediments within interval 4526m to 4554m have a very favourable oil source character, which appears to be genetically related to the crude oils tested in the upper portion of the well.

### C. Crude Oil Characterisation

Two (2) production tests performed over well intervals 3756m to 3783m and 3911m to 3914m recovered crude oil, gas and water. Crude oil sample numners 2 to 10 were recovered in Test 1 (3911m to 3914m) and sample numbers 15 to 28 were obtained in Test 2(3756m to 3783m) (Table 5). Sample G was collected during Test II and believed to be a representative sample from the interval. Crude Oil Characterization (Cont)....

The oils recovered during these tests are medium to high gravity crudes, with some of the higher gravities representing a mixture of very light oil and condensate ( $\#_{s}$  6-8 and  $\#_{s}$  15 - 16). These light oils/condensates from Test 2, appear to have been recovered during the early stages of the test, which suggests that there was some natural liquid chromatographic separation of the oils within the reservoir. As the test proceeded, a medium gravity, waxy crude was produced. Both condensates and crude oils have low sulphur contents and high pour points, indicating their high wax content. Waxy crude oils are generally sourced from terrestrial organic matter, which is very prevalent in the sediments penetrated by this well.

Both the high and medium gravity crudes are deficient in wet gas  $(C_2 - C_4)$  and gasoline range  $(C_5 - C_7)$  hydrocarbon. This may be a function of the source rocks not being mature enough to generate the more volatile hydrocarbons from the terrestrial organic matter within the sediments.

As mentioned in the previous section on Hydrocarbon Source Characterisation, the samples between 4526m to 4554m have very favourable hydrocarbon source characteristics, and bear a resemblence to some of the oils produced from this well, in particular to samples 26 and G (Figures 6C and 6D). These similarities are shown in the  $C_{12}$ + saturate g.c. for rock samples from 4526m to 4554m, and the greater than  $C_{12}^+$  components from the whole oilg.c.s of oil samples 26 and G. The overall g.c. configurations for the  $C_{12}^+$ . components are similar and suggest that a genetic relationship exists between these rocks and crude oils.

In addition to the oils from this well, three oils from the Halibut A-1, Hapuku 1 and Tuna A-2 wells were analyzed by whole oil gas chromatography. These g.c's traces look similar and also resemble the whole oil g.c's of oil samples 26 and G from Volador No. 1 well. The gc configurations of the good oil source sediments from 4526m to 4554m in the Volador also resemble the gc's of these three crude oils, and indicate a possible genetic relationship between the crudes and the favourable source rocks encountered at the bottom of Volador No. 1 well. Before a crude oil-parent source rock genetic relationship can be established between the Gippsland crudes and the sediments penetrated at the bottom of this well, a more detailed analytical program, utilising GC-MS should be undertaken on these rock and oil samples.

### D. Gas Characterization

The three (3) gas cylinder samples were analyzed (Table 9) to contain predominantly  $C_1$  methane, with secondary amounts of  $CO_2$ .

С.

### D. Gas Characterization (Cont)....

Carbon isotopic analysis (Table 10) indicates a thermal origin for the methane, as opposed to a biogenic origin, and an organic origin for the CO<sub>2</sub>, as opposed to a nonorganic origin (i.e. carbonate breakdown). Consequently the methane gas reservoired in this well is apparently migrated out-of-place from very thermally mature sediments.

The interpretation parameters for carbon isotopic data is as follows:-

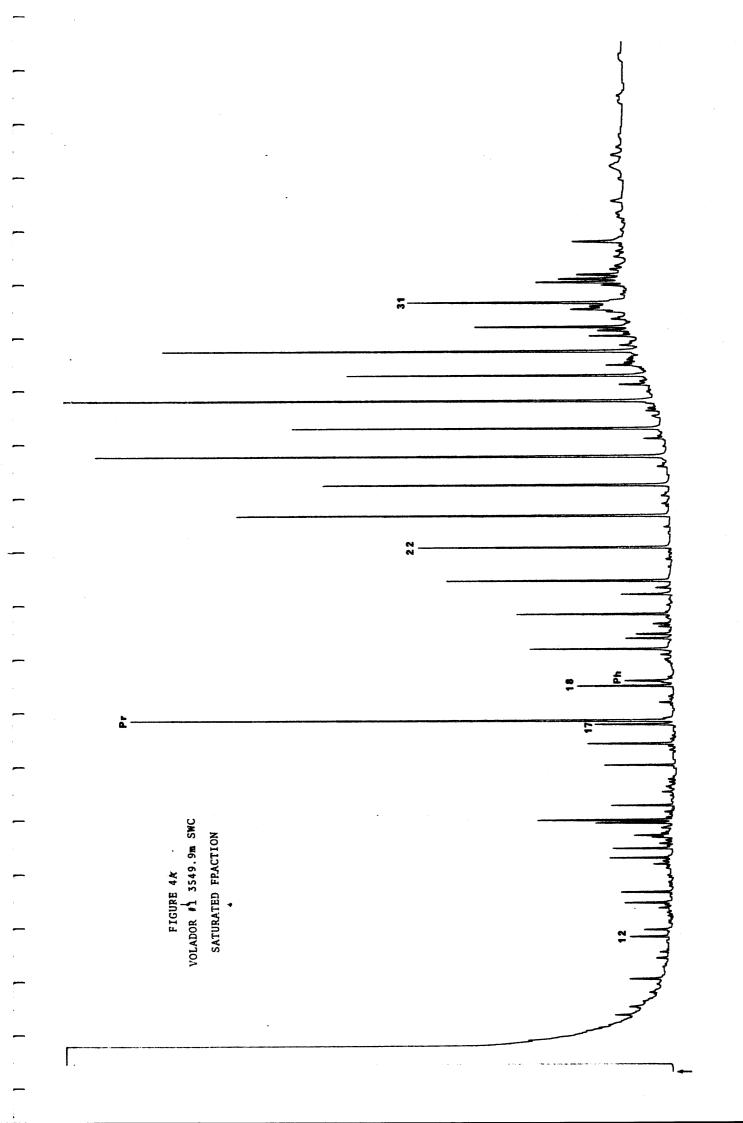
Methane

<u>δ13</u> C	Origin
-85 to -58	Biogenic
-58 to -40	Wetgas associated with oil
-40 to -25	Thermal

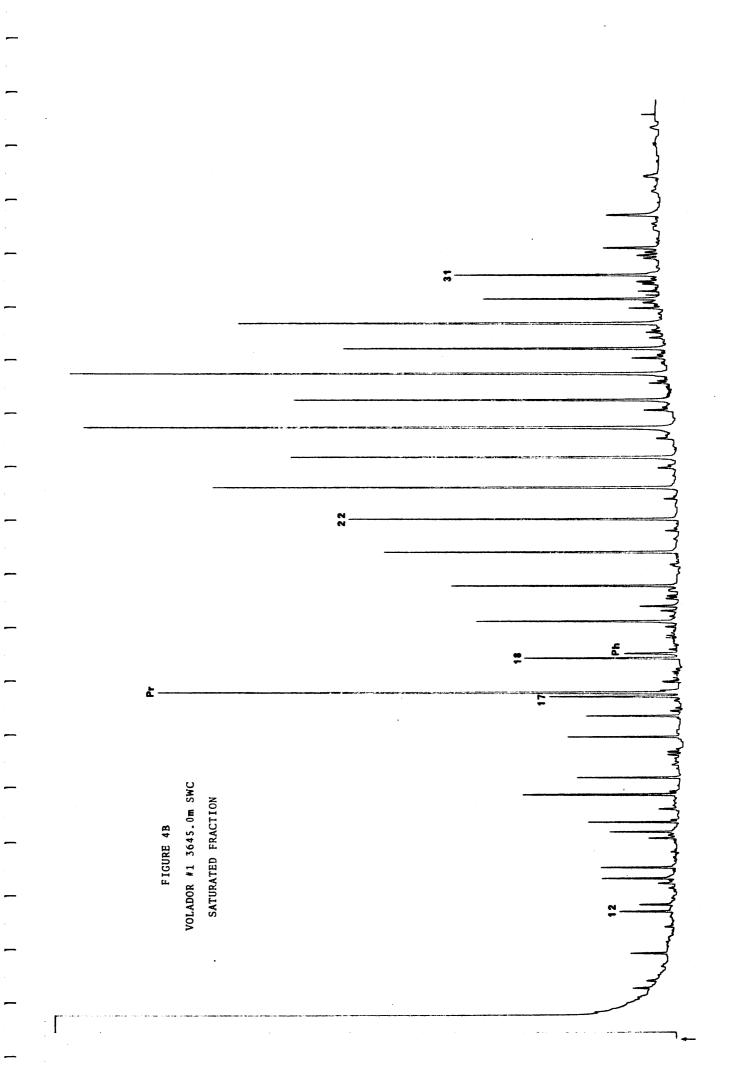
The CO<sub>2</sub> is probably sourced from the humic materials contained in these rocks based on the moderately negative  $\delta^{13}$ C values obtained from the gases.

### E. Water Analysis

Complete water analyses performed on water produced from the two (2) production tests carried out on this well indicate that the waters produced from Test 1 (3911m to 3914m) are considerably richer in salt than the waters produced in Test 2(3756m to 3783m).

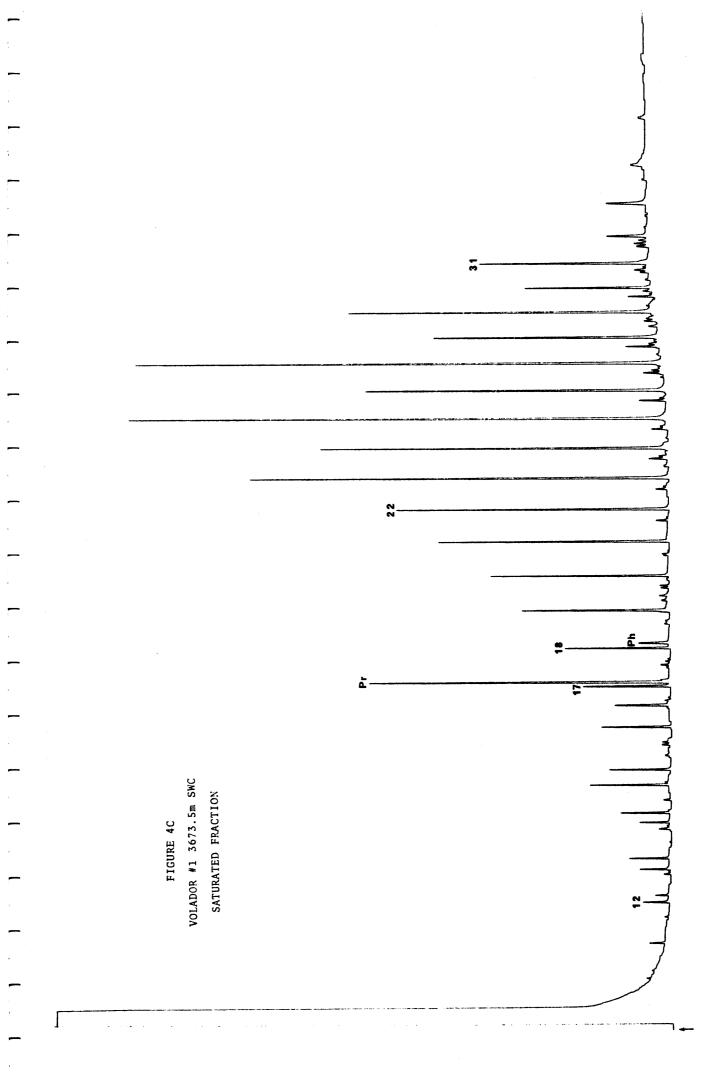


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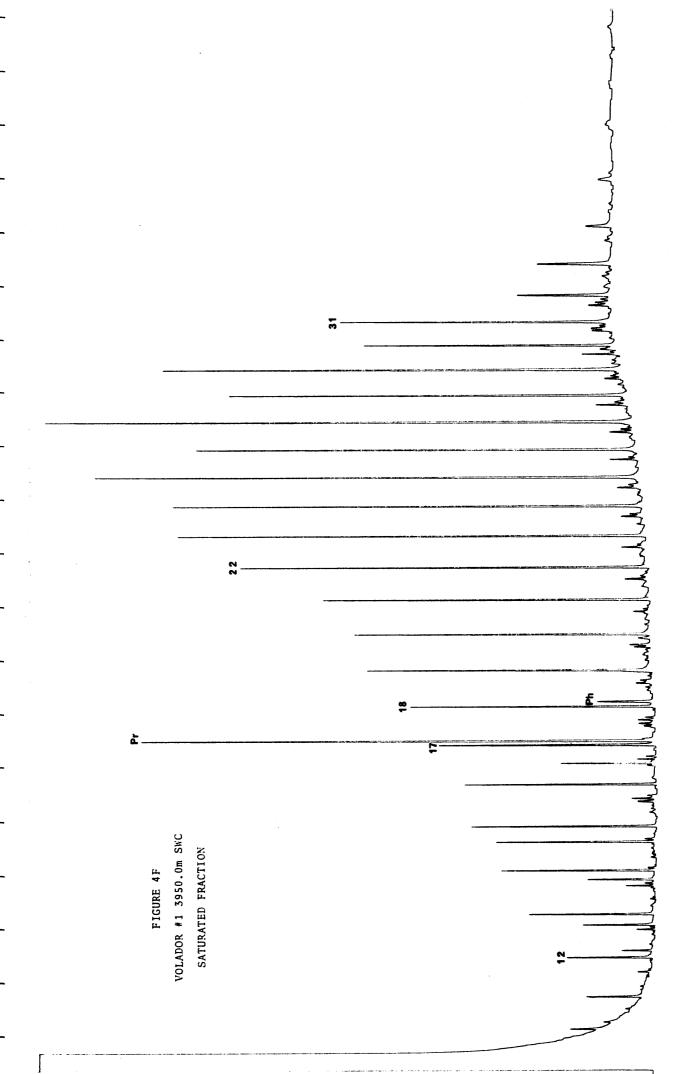


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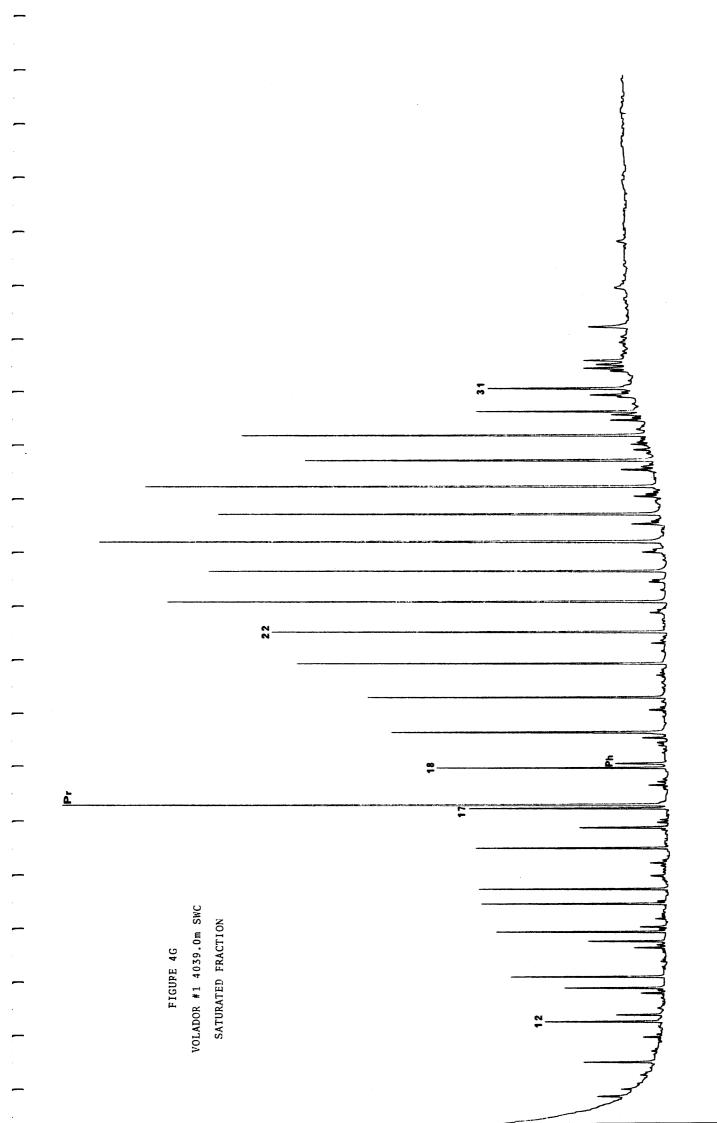
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	FIGURE 4E VOLADOR #1 3820.3m SWC SATURATED FRACTION	
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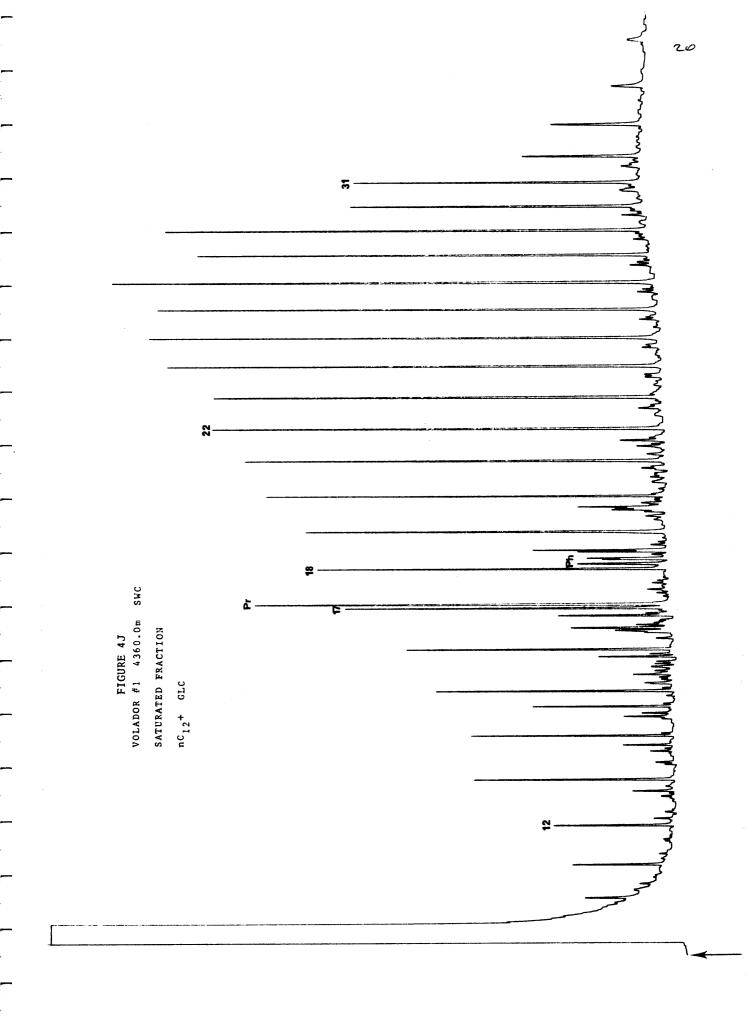
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FIGURE 4H VOLADOR #1 4145.3m SWC SATURATED FRACTION nC <sub>12</sub> + GLC	82
FIC VOLADOR #1 SATURATED 1 nC <sub>12</sub> + GLC	

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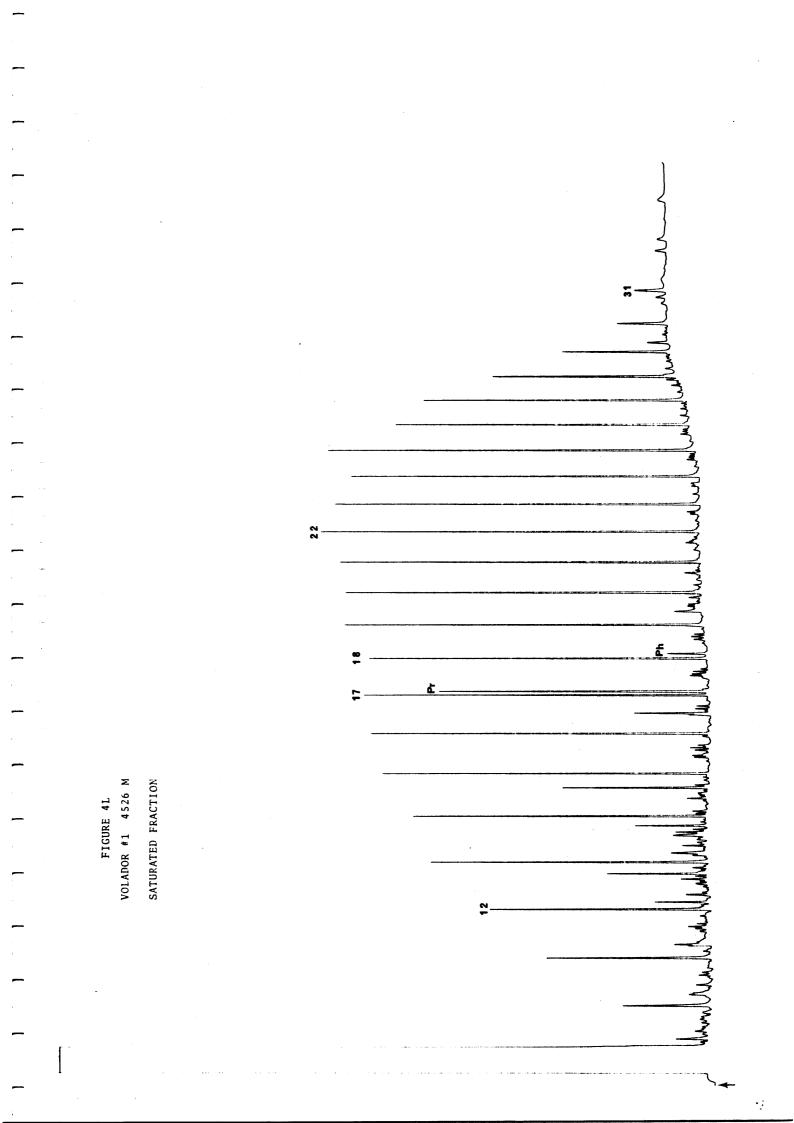
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FIGURE 4I VOLADOR #1 4265.0m SATURATED FRACTION nC <sub>12</sub> + GLC	
FI VOLADOR #1 SATURATED F nC <sub>1</sub> 2+ GLC	
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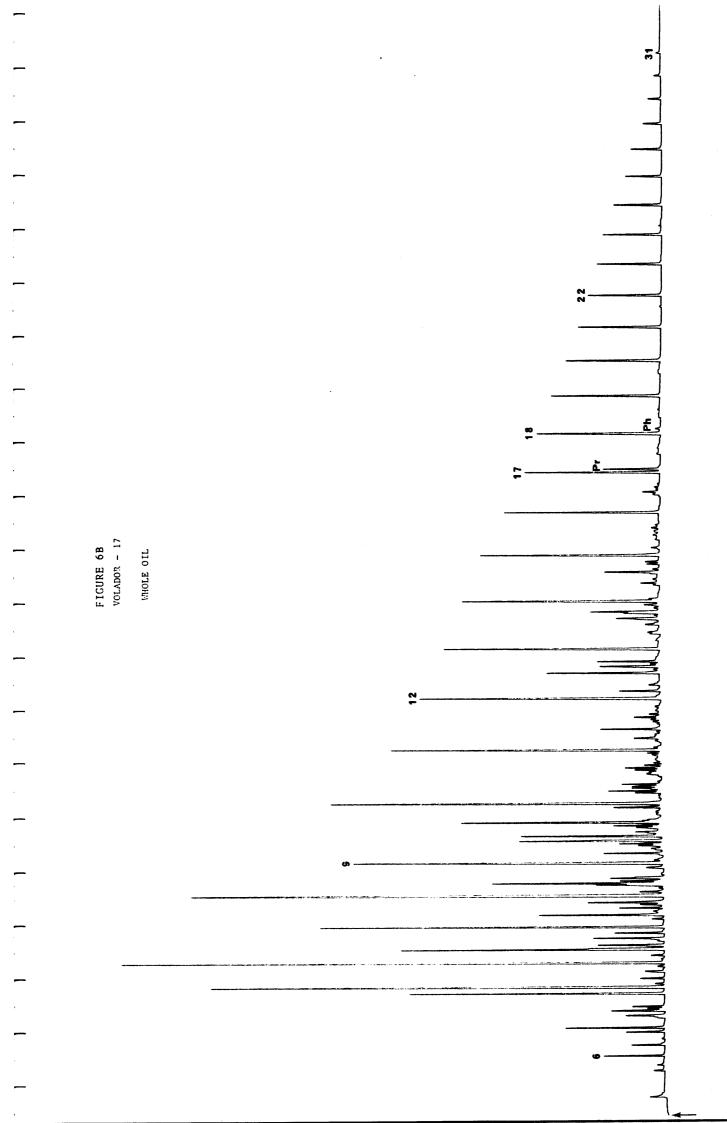
5 Š 22 £ 17<sub>Pr</sub> FIGURE 4N VOLATOR #1 4554 M SATURATED FRACTION 2

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FIGURE 7A Halibut a-1 Vellhead Sayfle										
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FIGURE 7B HAPUKU #1 FIT #5 WHOLE OIL	
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31 2 TUNA A-2 WELLHEAD SAMPLE FIGURE 7C 1996 – 2005m LIO BIOR 22 Ź đ. à 12

### ROCK-EVAL PYROLYSIS DATA (one run)

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NELLNAME = V	OLADOR #1							DATE OF	JOB = APRII	L, 1983	
DEPTH (m)	THAX	<b>S</b> 1	<b>S</b> 2	<b>S</b> 3	<b>51+52</b>	S2/S3	PI	PC	TOC	HI	01
3549.9	423	15.64	213.26	2.67	228.90	79.87	0.07	19.00	64.30	331	- 4
3645.0	427	2.09	51.45	2.63	53.54	19.56	0.04	4.44	17.00	302	15
3673.5	428	3.29	91.95	2.98	95.24	30.86	0.03	7.90	31.30	293	9
3691.5	429	1.60	38.12	2.50	39.72	15.25	0.04	3.30	13.90	274	17
3799.0	433	11.28	180.76	7.94	192.04	22.77	0.06	15.94	68.80	262	11
3820.3	433	13.04	214.49	5.21	227.53	41.17	0.06	18.88	73.50	291	7
3920.0	437	7.85	184.76	9.52	192.61	19.41	0.04	15.99	73.90	250	12
3950.0	441	1.27	14.72	2.63	15.99	5.60	0.08	1.33	8.60	171	30
4039.0	437	11.11	178.47	5.55	189.58	32.16	0.06	15.74	80.90	220	6
4145.3	443	0.89	4.88	1.69	5.77	2.89	0.15	0.48	3.22	151	52
4265.0	446	0.80	2.88	1.04	3.68	2.77	0.22	0.31	2.34	123	44
4360.0	444	0.74	2.07	0.85	2.81	2.44	0.26	0.23	1.80	115	47
4383.0	447	0.90	2.76	1.91	3.66	1.45	0.25	0.30	2.31	119	82
4526.0	451	8.65	52.88	2.11	61.53	25.06	0.14	5.11	21.71	243	9
4536.0	451	3.56	20.68	1.81	24.24	11.43	0.15	2.01	10.83	190	16
4554.0	448	0.87	4.48	3.39	5.35	1.32	0.16	0.44	4.28	104	79

TMAX	2	Max. temperature S2
51+S2	=	Potential yield
PC	z	Pyrolysable carbon
01	=	Oxygen Index

= Volatile hydrocarbons (HC) **S**1

= Organic carbon dioxide **S**3

TOC = Total organic carbon

≖ no data nd

# HC generating potential # Production index S2

PI

HI = Hydrogen index

### Summary of Extraction and Liquid Chromatography

Date of Job: APRIL, 1983

Date of Job: APRIL, 1983

A. Concentrations of Extracted Naterial

Wellname: VOLADOR #1

				Hyd	rocarbons-		N	ionhydrocarboi	ns
	Weight of	Total	Loss on			HC			NonHC
	Rock Extd.	Extract	Coluen	Saturates	Aromatics	Total	NSO's	Asphaltenes	Total
Depth(m)	(grams)	(ppm)	(ppm)	(ppm)	(ppe)	(ppm)	(ppe)	(ppa)	(ppm)
3549.9	9.4	11766.0	2010.6	2414.9	5478.7	7893.6	957.4	904.3	1861.7
3645.0	15.8	2613.9	278.5	696.2	1019.0	1715.2	303.8	316.5	620.3
3673.5	18.8	4319.1	287.2	1893.6	1457.4	3351.1	292.6	388.3	680.9
3691.5	20.4	1995.1	647.1	500.0	529.4	1029.4	225.5	93.1	318.6
3820.3	8.4	12250.0	1369.0	3392.9	4773.8	8166.7	1131.0	1583.3	2714.3
3950.0	6.7	3134.3	522.4	671.6	820.9	1492.5	776.1	343.3	1119.4
4039.0	8.4	9369.0	1738.1	2547.6	3023.8	5571.4	797.6	1261.9	2059.5
4145.3	10.9	1917.4	568.8	532.1	211.0	743.1	275.2	330.3	605.5
4265.0	9.1	1681.3	703.3	318.7	142.9	461.5	274.7	241.8	516.5
4360.0	3.9	1615.4	nd	nd	nd	nd	nd	nd	nd
4383.0	11.3	2123.9	539.8	654.9	336.3	991.2	398.2	194.7	592.9
4526.0	4.8	19187.5	3770.8	2375.0	1958.3	4333.3	5520.8	5562.5	11083.3
4536.0	25.0	7732.0	3395.9	1155.4	883.9	2039.3	1614.7	682.1	2296.8
4554.0	13.9	2244.6	798.6	402.9	273.4	676.3	518.0	251.8	769.8

### TABLE 2

Summary of Extraction and Liquid Chromatography

Wellname: VOLADOR #1

B. Compositional Data ASPH SAT HC ----Hydrocarbons---------Nonhydrocarbons-----EOM(eq) SAT (mg) ZHC's ZNSO's ZASPH. ZNon HC's TOC (q) AROM NSO Non HC Depth(m) ZSAT. ZARON. TOC (q) 3549.9 80.9 9.8 19.1 18.3 3.8 .44 .94 4.2 24.8 56.2 9.3 13.6 2.8 3645.0 29.8 73.4 26.6 15.4 4.1 . 68 1.04 43.6 13.0 1.30 1.33 4.9 16.9 13.8 6.0 3673.5 47.0 36.1 83.1 7.3 9.6 .94 3.2 3691.5 37.1 39.3 76.4 16.7 6.9 23.6 14.4 3.6 .41 3820.3 43.9 75.1 10.4 14.6 24.9 16.7 4.6 .71 1.40 3.0 31.2 31.4 57.1 29.7 42.9 36.4 7.8 .82 . 44 1.3 3950.0 25.7 13.1 27.0 11.6 3.1 .84 1.58 2.7 4039.0 33.4 39.6 73.0 10.5 16.5 39.5 24.5 44.9 59.5 16.5 2.52 1.20 1.2 4145.3 15.6 55.1 20.4 4265.0 14.6 47.2 28.1 24.7 52.8 71.9 13.6 2.23 .88 .9 32.6 4360.0 nd nd nd nd nd 89.7 nd nd nd nd nd 25.1 12.3 37.4 91.9 28.3 1.95 .49 1.7 4383.0 41.3 21.2 62.6 10.9 1.21 1.01 .4 28.1 35.8 71.9 88.4 4526.0 15.4 12.7 36.1 .9 1.31 .42 4536.0 20.4 47.0 37.2 15.7 53.0 71.4 10.7 26.6 .9 53.2 52.4 9.4 1.47 .49 4554.0 27.9 18.9 46.8 35.8 17.4

Summary of Gas Chromatography Data

A. Alkane Compositional Data

Wellname: VOLADOR #1

,

Depth(m)	Prist./Phyt.	Prist./n-C17	Phyt:/n-C18	CPI(1)	CPI (2)	(C21+C22)/(C28+C29)
3549.9	10.47	8.90	.71	1.77	1.71	. 58
3645.0	8.18	5.00	.54	1.63	1.60	80
3673.5	10.05	4.37	. 35	1.77	1.74	.87
3691.5	8.21	3.32	.37	1.67	1.63	.83
3820.3	14.00	4.41	. 30	1.51	1.48	1.10
3950.0	9.17	3.11	.29	1.26	1.21	.80
4039.0	12.06	3.94	. 30	1.31	1.26	. 98
4145.3	7.23	1.79	.24	1.24	1.18	.99
4265.0	6.37	1.38	.20	1.19	1.13	1.08
4360.0	4.68	1.94	. 36	1.11	1.06	. 89
4383.0	4.83	1.50	.29	1.13	1.08	.85
4526.0	6.79	.96	.14	1.07	1.05	2.03
4536.0	6.41	1.00	.15	1.06	1.05	1.91
4554.0	5.94	1.19	.20	1.11	1.07	1.33

TABLE 3

Summary of Gas Chromatography Data

Wellname: VOLADOR #1

Date of Job:APRIL, 1983

B. n-Alkane Distributions

DEPTH(a)	nC12	nC13	nC14	nC15	nC16	nC17	iC19	nC18	i C20	nC19	nC20	nC21	nC22	nC23	nC24	nC25	nC26	nC27	nC28	nC29	nC30	nC31
3549.9	.7	.9	1.0	1.1	1.2	1.3	11.9	1.6	1.1	2.6	2.7	3.8	4.5	7.6	6.3	10.7	6.7	13.1	5.6	8.7	3.0	4.0
3645.0	.7	1.1	1.2	1.4	1.6	2.0	10.0	2.3	1.2	3.2	3.5	4.5	5.2	8.4	6.7	11.3	6.6	10.7	5.2	7.0	2.8	3.3
3673.5	.5	.7	.9	1.2	1.3	1.7	7.4	2.1	.7	3.0	3.4	4.6	5.7	9.2	7.3	12.4	6.7	12.6	4.9	6.9	2.8	3.7
3691.5	1.1	1.5	1.7	1.9	2.0	2.3	7.6	2.5	.9	3.4	3.5	4.6	5.3	8.4	6.8	11.4	6.5	10.7	5.0	6.9	2.7	3.3
3820.3	1.3	2.0	2.3	2.4	2.3	2.7	11.8	2.8	.8	3.6	4.0	4.9	5.7	7.9	6.8	9.8	6.1	8.6	4.2	5.5	2.1	2.2
3950.0	1.1	1.6	2.0	2.4	2.5	2.8	8.7	3.3	.9	4.0	4.1	4.7	5.4	6.3	6.6	8.0	6.7	8.6	5.8	6.8	3.7	4.2
4039.0	1.6	2.0	2.3	2.5	2.5	2.8	11.1	3.1	.9	3.9	4.1	5.1	5.7	7.0	6.7	8.3	6.7	8.2	4.9	6.1	2.4	2.1
4145.3	1.9	2.7	2.8	2.9	3.0	3.4	6.0	3.5	.8	4.2	4.3	5.1	5.9	6.8	7.0	8.2	6.8	8.1	5.2	5.8	2.6	2.9
4265.0	1.9	2.3	2.7	3.0	3.1	3.5	4.9	3.8	.8	4.5	4.8	5.5	6.2	7.0	7.3	7.9	7.0	7.7	5.2	5.7	2,5	2.6
4360.0	1.3	2.1	2.2	2.6	3.0	3.5	6.8	4.0	1.4	4.6	4.7	5.2	5.5	6.2	6.5	6.9	6.6	7.3	6.0	6.1	3.7	3.9
4383.0	1.8	2.3	2.5	2.8	2.9	3.2	4.7	3.4	1.0	4.1	4.3	4.9	5.6	6.5	7.0	8.0	7.4	8.1	6.3	6.2	3.4	3.5
4526.0	3.8	4.5	4.9	5.1	5.2	5.4	5.1	5.4	.8	5.8	5.7	5.9	6.1	6.1	6.0	6.0	5.1	4.9	3.5	2.4	1.3	1.0
4536.0	4.1	4.9	5.4	5.5	5.4	5.5	5.5	5.7	.9	5.9	5.3	5.5	5.6	5.7	5.6	5.7	4.9	4.7	3.3	2.5	1.4	.9
4554.0	3.1	3.3	3.4	3.5	3.7	4.3	5.1	4.3	.9	4.9	4.8	5.4	6.3	6.7	6.9	7.3	6.5	6.8	4.8	4.0	2.1	2.0

na = not applicable nd = no data

### RESULTS OF VITRINITE REFLECTANCE

Depth (m)	R Max	Range
3549.9	0.61	0.50 - 0.68
3645	0.71	0.60 - 0.80
3673.5	0.72	0.66 - 0.86
3691.5	0.82	0.63 - 0.93
3799	0.75	0.66 - 0.89
3820.3	0.76	0.67 - 0.88
3920	0.85	0.75 - 0.93
3950	0.94	0.75 - 1.05
4039	0.89	0.82 - 0.96
4152	0.85	0.72 - 0.95
4170	0.88	0.68 - 1.03
4191	0.89	0.75 - 1.01
4218	0.85	0.72 - 0.98
4264	0.83	0.70 - 0.94
4372	0.88	0.75 - 0.99
4526	0.90	0.81 - 0.98
4536	0.89	0.77 - 1.02
4554	0.89	0.74 - 1.01

\* \* \* \* \* \* \* \* \* \* \* \* \* \* \*

PHYSICAL PROPERTY DATA - VOLADOR #1 OILS

1

Sample	API Gravity(60°F)	<pre>% Sulphur(w/w)</pre>	<pre>% Sulphur(w/w) Viscosity(25°C) Viscosity(60°C)</pre>	Viscosity(60°C)	Pour Viscosity(100°C) Point (°C)	Pour Point (°C)
2	39.6	I	T	T	8	
9	42.8	ı	ı		ı	ı
7	42.8	·	ı	ł	ı	1
8	41.1	0.047	16.06	1.90	ı	+27
10	39.6	ı	ł	ı	ı	I
15	43.6	0.067	1.82	1.17	·	+12
17	43.6	I	ı	1	ı	ı
18	36.8	ı	ı	•	ı	ı
20	37.2	J	ı	ı	ı	ı
21	36.3	•	·	ı	ı	ı
22	36.3	·	ı	ı	ı	ı
23	36.6	ı	•	ı		ľ
24	34.1		ı	•	ı	ı
25	33.3	·	·	ı	ı	ı
26	33.7	0.069	l	5.00	2.63	+33
28	34.5	ı	ı	•	1	ı
IJ	37.2	0.062	ı	3.75	2.12	+36

N.B. Units for viscosity are centistokes.

## COMPOSITIONAL DATA

ł

	21+22/28+29	•¢
	21+22	1.6
	CPI(2)	1.06
	CPI (1)	1.08
	ARDM/SAT	0.18
	PHYT/NC18	
	PRIST/NC17	.73
	PRIST/PHYT	6.79
	ZNSO	2.6
	ZARON	15.1
	ZSAT	82.3
	OILNAKE	VOLADOR EX SAMPLE 8
ŗ	i	ł

# N-ALKANE DISTRIBUTIONS

- OILNARE - Volador ex Sanple 8 ŧ

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CN12 CN13 CN14 CN15 CN16 CN17 CN18 CN19 CN20 CN21 CN22 CN23 CN24 CN25 CN26 CN27 CN28 CN29 CN30 CN31 8.1 7.7 7.4 6.7 6.0 5.6 5.3 5.2 4.9 4.8 5.0 5.1 5.1 5.3 4.6 4.7 3.4 2.6 1.4 1.0

			and the second secon	
Carbon Number	Sample G	Sample 8	Sample 17	Sample 26
1 - 3	0.39	0.04	<0.01	<0.01
4	0.98	0.24	0.01	0.04
5	1.7	1.1	0.02	0.54
6	3.9	5.2	2.1	1.5
7	7.0	13.9	14.4	3.9
8	8.3	18.0	23.4	6.6
9	4.6	9.9	12.7	5.5
10	3.6	6.6	7.7	5.0
11	2.9	5.4	5.9	4.9
12	3.2	5.7	6.4	5.9
13	3.2	4.6	5.4	5.9
14	3.3	3.9	4.1	5.7
15	3.4	6.5	3.3	5.0
16	3.0	3.5	2.5	4.6
17	3.9	3.0	2.6	5.3
18	3.4	1.8	1.8	3.7
19	3.6	1.5	1.4	3.3
20	4.0	1.4	1.1	3.4
21	4.2	1.1	1.0	3.3
22	4.5	1.1	0.80	3.5
23	4.6	1.0	0.72	3.5
24	4.7	1.0	0.64	3.6
25	4.6	0.90	0.53	3.6
26	3.6	0.74	0.39	3.2
27	3.6	0.72	0.37	3.1
28	2.3	0.48	0.25	2.1
29	1.8	0.34	0.21	1.6
30	1.0	0.18	0.11	1.0
31	0.84	0.13	0.08	0.72

N.B. All values expressed as weight %

TABLE7

COMPOSITIONAL DATA - VOLADOR #1\_OILS

			Tura 1 2
Carbon Number	Halibut A-1	Hapuku #1	Tuna A-2
1 - 3	<0.01	<0.01	<0.01
4	0.14	0:05	<0.01
5	0.34	0.16	<0.01
6	2.5	2.4	0.12
7	6.2	6.0	0.27
8	9.2	7.9	1.6
9	5.8	5.8	3.3
10	4.7	5.1	3.4
11	4.2	5.0	4.4
12	5.1	6.7	7.1
13	5.7	7.3	9.5
14	5.2	7.5	9.0
15	4.7	5.4	6.6
16	4.5	5.0	10.4
17	6.5	5.6	6.3
18	4.1	3.9	5.5
19	4.2	3.4	5.1
20	4.0	3.0	4.6
21	3.4	2.9	4.2
22	3.3	2.7	3.8
23	3.1	2.6	3.4
24	2.8	2.4	3.0
25	2.6	2.4	2.5
26	2.0	1.9	2.1
27	2.0	1.8	1.4
28	1.4	1.2	1.1
29	0.96	0.79	0.55
30	0.56	0.48	0.41
31	0.43	0.31	0.44

# COMPOSITIONAL DATA - GIPPSLAND OILS

TABLE 8

N.B. All values expressed as weight %.

40

### TABLE 9

### GAS ANALYSIS DATA - VOLADOR #1

Component	Cylinder A 4974	Cylinder 5283	Cylinder 13764
N2	2.90	2.52	1.98
$0_2^2 + Ar$	0.22	0.62	0.49
cõ,	36.0	15.2	14.2
Methane	46.0	72.0	71.6
Ethane	6.1	5.6	8.0
Propane	5.5	2.4	2.7
iso-Butane	0.94	0.35	0.29
n-Butane	1.28	0.58	0.43
neo-Pentane	0.0025	0.0016	0.0014
iso-Pentane	0.15	0.16	0.107
n-Pentane	0.013	0.174	0.098
hexanes	0.0062	0.135	0.052
heptanes	0.0044	0.126	0.046
octanes	0.00607	0.092	0.023

\*\*\*\*\*

# N.B. Values expressed as volume $\$

# TABLE 10

## Carbon Isotope Data

Bottle No.	$\delta^{13}$ C Hydrocarbons	$\delta^{13}$ C Carbon Dioxide
A 4974	- 30.2	-21.2
13764	- 32.2	-12.7
5283	- 34.8	-19.2

N.B. Values expressed relative to PDB limestone.

\* \* \* \* \* \* \* \* \* \* \* \*

TABLE 11A
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ANALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

### CERTIFICATE OF

For: Shell Development 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 17.05.83

ANALYSIS

Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

1

230800 147700

Method of Analysis:

Sample No.

7.46

Chemical Data

pH Conductivity(u siemens/cm) T.F.R. (calculated)

		mg/l	m equi∨/l
Sodium	Na+	49000	2131
Potassium	K+	405	10.36
Calcium	Ca++	600	29.94
Magnesium	Mg++	1325	109
Soluble Iron	Fe	1.75	-
Chloride	C 1-	76680	2160
Carbonate	C03	<0.3	
Bi-Carbonate	HC03-	457.5	7.5
Sulphate	S04	3265	68.02
Nitrate	N03-	1.439	0.0232

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

ANALABS ANALYTICAL CHEMISTS CERTIFICATE Shell Development For: 140 St George's Terrace Perth W.A. 6000 Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis. Sample No. Method of Analysis:  $\mathbf{2}$ Chemical Data 7.58 юH Conductivity(u siemens/cm) 235300 150600 T.F.R. (calculated) mg/153000 Na+ Sodium 375 Potassium K+ 600 Ca++ Calcium 1350 Magnesium Mg++ 1.5 Soluble Iron Fe 83210 Chloride C1-<0.3 Carbonate C03---433.1 Bi-Carbonate HCO3-3000 Sulphate 804--2.773 Nitrate N03-

52 Murnay Road Welshpool W.A. 6106 Tel: 458 7999

AHALYSIS

m equiv/l

2305

2344

9.591

29.94

\_

7.1

0.0447

62.5

111.1

Our ref: 1000.01.27272

Your ref: 17.05.83 Date:

A.P.T.C., A.R.A.C.I., A.A.I.M.M. Analyst: M.A. CHAPMAN

Analytical Chemist

### ANALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

# CERTIFICATE OF ANALYSIS

For: Shell Development 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 17.05.83

Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

Method of Analysis:		Sample No.	
Chemical Data	B.	3	
рН		7.4	
Conductivity	(u siemens/cm)	224000	
T.F.R. (calco	ulated)	143400	
		mg/ 1	m equiv $21$
Sodium	Na+	51000	2218
Potassium	K+	405	10.36
Calcium	Ca++	600	29.94
Magnesium	Mg++	1350	111.1
Soluble Iron	Fe	1.6	-
Chloride	C 1-	80660	2272
Carbonate	C03	<0.3	· _
Bi-Carbonate	HC03-	457.5	7.5
Sulphate	S04	3105	64.69
Nitrate	N03-	1.661	0.0268

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

TABLE 11D

### ANALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

CERTIFICATE OF ANALYSIS

For: Shell Developement 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 6.05.83

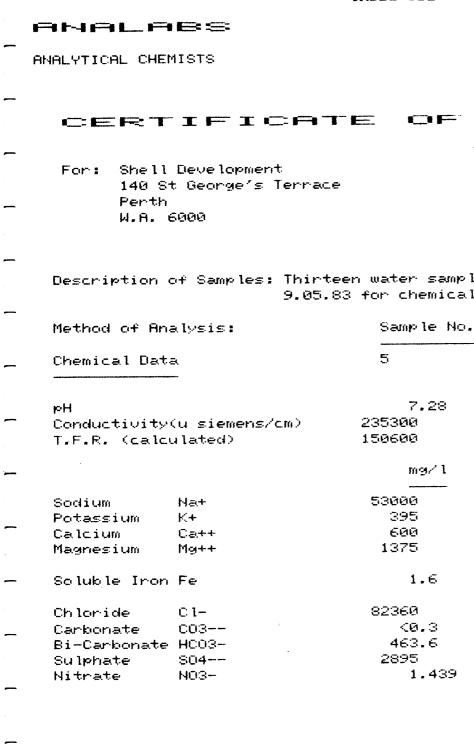
Description of Samples: Five water samples were received on the 20.04.83 for chemical analysis.

	Method of Analysis:		Sample No.	
_	Chemical Data		No4	
_	рН Conductivity	(u siemens/cm)	6.55 245300	
	T.F.R. (calc		157000	
-			mg/1	m equiv/l
	Sodium	Na+	55800	2427
_	Potassium	K+	440	11.25
	Calcium	Ca++	470	23.45
	Magnesium	Mg++	1300	107
-	Soluble Iron	Fe	47	
	Chloride	C 1-	86340	2432
_	Carbonate	C03	<0.3	-
	Bi-Carbonate	HC03-	1659	27.2
	Sulphate	S04	3050	63.54
_	Nitrate	N03-	3.062	0.0494

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

TA	Bl	LΕ	1	1	Ε
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52 Munnay Road Welshpool W.A. 6106 Tel: 458 7999

ANALYSIS 

m equiv/l

2305

2320

10.1

113.2

29.94

.....

7.6

60.31

0.0232

Our ref: 1000.01.27272

Your ref: 17.05.83 Date:

Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

7.28

mg/1

1.6

1.439

A.P.T.C., A.R.A.C.I., A.A.I.M.M. Analyst: M.A. CHAPMAN

Analytical Chemist

TABLE 11F

### ANALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

### CERTIFICATE OF ANALYSIS

For: Shell Development 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 17.05.83

Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

Method of Ar	alysis:	Sample No.	
Chemical Dat		6	
юН		6.29	
Conductivity	(u siemens/cm)	264000	
T.F.R. (cal		169000	
		mg/1	m equi∪/l
Sodium	Na+	57000	2479
Potassium	K+	395	10.1
Calcium	Ca++	525	26.2
Magnesium	Mg++	1350	111.1
Soluble Iron	) Fe	135	-
Chloride	C 1-	88320	2488
Carbonate	03	<0.3	
Bi-Carbonate		201.3	3.3
Sulphate	S04	3105	64.69
Nitrate	N03-	13.29	0.214

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.L., A.A.I.M.M.

Analytical Chemist

TABLE 11G

### ANALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

### CERTIFICATE OF ANALYSIS

For: Shell Development 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 17.05.83

Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

7

Method of Analysis:

Sample No.

Chemical Data

pH Conductivity(u siemens/cm) T.F.R. (calculated)		6.5 256800 164300	
		mg∕ 1	m equiv/l
Sodium	Na+	57000	2479
Potassium	K+	385	9.847
Calcium	Ca++	500	24.95
Magnesium	Mg++	1325	109
Soluble Iron	Fe	140	-
Chloride	C 1-	87190	2456
Carbonate	C03	<0.3	
Bi-Carbonate	HC03-	286.7	4.7
Sulphate	S04	3790	78.96
Nitrate	N03-	14.39	0.232

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

### TABLE 11H

### ANALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

### CERTIFICATE OF ANALYSIS

For: Shell Development 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 17.05.83

Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

Method of An	alysis:	Sample No.	
Chemical Data	a -	8	
рН		6.17	
Conductivity	(u siemens/cm)	257000	
T.F.R. (calc	ulated)	164500	
		m97-1	m equival
Sodium	Na+	57000	2479
Potassium	K+	385	9.847
Calcium	Ca++	500	24.95
Magnesium	Mg++	1325	109
Soluble Iron	Fe	145	
Chloride	C 1-	90880	2560
Carbonate	C03	<0.3	
Bi-Carbonate	HC03-	317.2	5.2
Sulphate	S04	3895	81.15
Nitrate	N03-	17.71	0.2857

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

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### ANALABS

Perth

W.A. 6000

Shell Development

140 St George's Terrace

ANALYTICAL CHEMISTS

For:

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

### CERTIFICATE OF ANALYSIS

Our ref: 1000.01.27272

Your ref: Date: 17.05.83

Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

Method of Analysis: Sample No. 9 Chemical Data 6.31 pН 255500 Conductivity(u siemens/cm) 163500 T.F.R. (calculated) mg/1m = equiv/159500 2588 Sodium Ma+ 395 10.1 Potassium K+ 475 23.7 Ca++ Calcium 1300 107 Mg++ Magnesium ..... 100 Soluble Iron Fe 2544 C 1-90310 Chloride <0.3 Carbonate CO3--5.7 347.7 Bi-Carbonate HCO3-68.02 3265 Sulphate \$04--17.71 0.2857 N03-Nitrate

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

Signature			TABLE 11J		
Tel: 458 7999 Tel: 458 7999 CERTIFICATE OF AMAL'SIS For: Shell Development 140 St George's Terrace Perth W.R. 6000 Description of Samples: Thirteen water samples were received on the 9.85.83 for chemical analysis. Method of Analysis: Sample No. Chemical Data PH 6.45 Conductivity(u siemens/cm) 254000 T.F.R. (calculated) 162600 Magnesium Na+ Sodium Na+ Sodium Na+ Sodium Na+ Sodium Ca++ 450 22.46 Magnesium Mg++ 1225 100.8 Soluble Iron Fe 49 - Chloride C1- Chemical C03- Bi-Carbonate C03- Soluble Iron Fe 49 - Chloride C1- Calculated) Calculated 512 Calculated 512 Carbonate C03- Soluble Iron Fe 49 - Chloride C1- Carbonate C03- Soluble Iron Fe 512.4 Soluble Iron Fe 512.4 Calculated 74.58	AHALA				
For:Shell Development 140 St George's Terrace Perth W.R. 6000Our ref:1000,01.27272 Your ref: Date:Description of Samples:Thirteen water samples were received on the 9,05.83 for chemical analysis.Date:17.05.83Description of Samples:Sample No.Chemical Data10PH6.45Conductivity(u siemens/cm)254000 162600T.F.R. (calculated)162600 $mg/1 = m = quiv/1$ SodiumNa+595002588 10.36CalciumCa++45010.36CalciumCa++45022.46MagnesiumMg++1225100.8Soluble Iron Fe49-ChlorideC1-898902532 2532CarbonateC03(0.3-Bi-CarbonateC03512.48.4SulphateS04356074.58	ANALYTICAL CHE	11878			
For: Shell Development 140 St George's Terrace Perth W.A. 6000 Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis. Method of Analysis: Sample No. Chemical Data PH 6.45 Conductivity(u siemens/cm) 7.F.R. (calculated) Method of Analysis: Method of Analysis: Method of Analysis: Method of Analysis: Method of Analysis: Sample No. Chemical Data PH 6.45 Conductivity(u siemens/cm) 7.F.R. (calculated) Method 59500 Meg/1 m equiv/1 Sodium Nat 59500 2588 Potassium K+ 405 10.36 Calcium Ca++ 456 22.46 Magnesium Mg++ 1225 100.8 Soluble Iron Fe Chloride C1- B9890 2532 Carbonate C03 Sil2.4 8.4 Sulphate S04 State 74.58					
The second of t			• <b>•</b> ••••••••••••••••••••••••••••••••••		
Perth W.R. 6000Your ref: Date:Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.Method of Analysis:Sample No.Chemical Data10pH6.45Conductivity(u siemens/cm)254000T.F.R. (calculated)162600mg/lm equiv/lSodiumNa+5950022588PotassiumK+40510.36CalciumCa++45022.46MagnesiumMg++1225100.8So luble Iron Fe49ChlorideC1-Bi-CarbonateC03ChorideC1-State5320Carbonate512.48.4358074.58			_	Our ref:	1000.01.27272
W.R. 6000Date:17.05.83Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.Nethod of Analysis:Sample No.Method of Analysis:Sample No.Chemical Data10pH6.45 Conductivity(u siemens/cm)254000T.F.R. (calculated)162600Method calculated)162600SodiumNa+ 4052588 10.36Calculated)1225SodiumNa+ 40522.46MagnesiumMg++1225Soluble Iron Fe49-ChlorideC1- 898962532 2532 CarbonateChlorideC1- 898962532 2532 - 63.3Bi-CarbonateC03 512.43.4 3580SulphateS043580Calculate512.4SulphateS04		t George's Terrac	E.	Your ref:	
9.05.83 for chemical analysis.Method of Analysis:Sample No.Chemical Data10pH6.45Conductivity(u siemens/cm)254000T.F.R. (calculated)162600mg/1m equiv/1SodiumNa+59500258810.36PotassiumK+40510.36CalciumCa++45022.46MagnesiumMg++1225100.8So luble Iron Fe49ChlorideC1-898902532CarbonateC0360.3-Bi-Carbonate403-512.48.4SulphateS04358074.58		5000			
Chemical Data       10         pH       6.45         Conductivity(u siemens/cm)       254000         T.F.R. (calculated)       162600         mg/l       m equiv/l         Sodium       Na+         Sodium       Na+         Sodium       K+         Quark       405         Potassium       K+         405       10.36         Calcium       Ca++         450       22.46         Magnesium       Mg++         1225       100.8         Soluble Iron Fe       49         Chloride       C1-         89890       2532         Carbonate       C03         Soluble Iron Fe       49         Bi-Carbonate HC03-       512.4         State       512.4	Description (	of Samples: Thirt 9.05.	een water sampl 83 for chemical	es were receive analysis.	d on the
pH       6.45         Conductivity(u siemens/cm)       254000         T.F.R. (calculated)       162600 $mg/1$ m equiv/l         Sodium       Na+         59500       2588         Potassium       K+         405       10.36         Calcium       Ca++         450       22.46         Magnesium       Mg++         1225       100.8         Soluble Iron Fe       49         Chloride       C1-         89890       2532         Carbonate       C03         8i-Carbonate       60.3         512.4       8.4         Sulphate       S04	Method of An	alysis:	Sample No.		
Conductivity(u siemens/cm)       254000         T.F.R. (calculated)       162600 $mg/1$ $m equiv/1$ Sodium       Na+         Sodium       Na+         9500       2588         Potassium       K+         405       10.36         Calcium       Ca++         450       22.46         Magnesium       Mg++         1225       100.8         Soluble Iron Fe       49         Chloride       C1-         89890       2532         Carbonate       C03         60.3       -         Bi-Carbonate       512.4         Soluphate       S04	Chemical Dat	<b>E</b> .	10		
T.F.R. (calculated)       162600         mg/l       m equiv/l         Sodium       Na+       59500       2588         Potassium       K+       405       10.36         Calcium       Ca++       450       22.46         Magnesium       Mg++       1225       100.8         Soluble Iron Fe       49       -         Chloride       C1-       89890       2532         Carbonate       C03       (0.3       -         Bi-Carbonate       HC03-       512.4       8.4         Sulphate       S04       3580       74.58	юΗ		6.45		
mg/1m equiv/1SodiumNa+595002588PotassiumK+40510.36CalciumCa++45022.46MagnesiumMg++1225100.8Soluble Iron Fe49-ChlorideC1-898902532CarbonateC03<0.3	Conductivity	(u siemens/cm)			
Sodium         Na+         59500         2588           Potassium         K+         405         10.36           Calcium         Ca++         450         22.46           Magnesium         Mg++         1225         100.8           Soluble Iron         Fe         49         -           Chloride         C1-         89890         2532           Carbonate         C03         <0.3	T.F.R. (calc	ulated)	162600		
Potassium       K+       405       10.36         Calcium       Ca++       450       22.46         Magnesium       Mg++       1225       100.8         Soluble Iron Fe       49       -         Chloride       C1-       89890       2532         Carbonate       C03       <0.3       -         Bi-Carbonate       HC03-       512.4       8.4         Sulphate       S04       3580       74.58			mg/1	m equiv/l	
Calcium       Ca++       450       22.46         Magnesium       Mg++       1225       100.8         Soluble Iron Fe       49       -         Chloride       C1-       89890       2532         Carbonate       C03       <0.3	Sodium	Na+	59500	2588	
Magnesium       Mg++       1225       100.8         Soluble Iron Fe       49       -         Chloride       C1-       89890       2532         Carbonate       C03       <0.3	Potassium	К+	405		
Soluble Iron Fe     49     -       Chloride     C1-     89890     2532       Carbonate     C03     <0.3	Calcium	Ca++	450		
Chloride     C1-     89890     2532       Carbonate     C03     <0.3	Magnesium	Mg++	1225	100.8	
Carbonate C03 <0.3 - Bi-Carbonate HC03- 512.4 8.4 Sulphate S04 3580 74.58	Soluble Iron	Fe	49	-	
Carbonate       C03       <0.3	Chloride	C1-	89890	2532	
Bi-Carbonate HCO3- 512.4 8.4 Sulphate SO4 3580 74.58					
Sulphate \$04 3580 74.58				8.4	
	•		13.29	0.2143	

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

TABLE 11K 52 Murray Road AHALABS Welshpool W.A. 6106 ANALYTICAL CHEMISTS Tel: 458 7999 CERTIFICATE OF ANALYSIS Our ref: 1000.01.27272 For: Shell Developement 140 St George's Terrace . Your ref: Perth Date: 6.05.83 W.A. 6000 Description of Samples: Five water samples were received on the 20.04.83 for chemical analysis. Sample No. Method of Analysis: No11 Chemical Data 7.72 pН Conductivity(u siemens/cm) 195000 T.F.R. (calculated) 124800 mg/1m equiv/l 1918 Sodium Na+ 44100 14.58 Potassium K+ 570 2.495 50 Calcium Ca++ 55.96 680 Magnesium Mg++ 2.9 Soluble Iron Fe -----64750 1824 Chloride C 1-<0.3 Carbonate C03--------2218 36.36 Bi-Carbonate HCO3-3200 66.67 Sulphate \$04---<0.05 Nitrate N03-.....

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

TABLE 11L

ANALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

CERTIFICATE OF ANALYSIS

For: Shell Development 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 17.05.83

Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

Method of Analysis: Sample No. 12 Chemical Data 8.58 pH 110500 Conductivity(u siemens/cm) 70720 T.F.R. (calculated) m equiv/l mg/1-----1044 24000 Sodium Na+ 12.15 475 Potassium K+ 6.487 130 Calcium Ca++ 23.04 280 Mg++ Magnesium 3.5 ----Soluble Iron Fe 968 34360 C1-Chloride 4.9 147 CO3--Carbonate 63.8 3892 Bi-Carbonate HCO3-51.77 2485 S04--Sulphate 0.1696 10.52 N03-Nitrate

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

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ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

### CERTIFICATE OF ANALYSIS

For: Shell Developement 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 6.05.83

Description of Samples: Five water samples were received on the 20.04.83 for chemical analysis.

Method of Analysis: Chemical Data		Sample No.	
		No14	
ρΗ		8.25	
Conductivity	(u siemens/cm)	45750	
T.F.R. (calcu	(lated)	29280	
		mg/ 1	m equiv∕l
Sodium	Na+	9825	427.4
	K+	360	9.207
Calcium	Ca++	30	1.497
Magnesium	M9++	60	4.937
Soluble Iron	Fe	0.25	-
Chloride	C 1-	14200	400
Carbonate	C03	<0.3	****
Bi-Carbonate	HC03-	2223	36.44
Sulphate	304	900	18.75
Nitrate	N03-	0.3062	0.004

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

TABLE 11N

ANALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

### CERTIFICATE OF ANALYSIS

For: Shell Developement 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 6.05.83

Description of Samples: Five water samples were received on the 20.04.83 for chemical analysis.

Method of Analysis: Chemical Data		Sample No.	
		No16	
рН		8.38	
Conductivity	(u siemens/cm)	45000	
T.F.R. (calcu	(lated)	28800	
		m⊛∕ 1	m equi∪/l
Sodium	Na+	9875	394.7
Potassium	<b>K</b> +	350	8.951
Calcium	Ca++	50	2.495
Magnesium	Mg++	55	4.526
Soluble Iron	Fe	0.05	-
Chloride	C 1-	13060	368
Carbonate	C03	13.2	0.44
Bi-Carbonate	HC03-	1952	32
Sulphate	S04	720	15
Nitrate	N03-	0.2165	0.003

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

TABLE	11	0
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### AHALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

#### HHHHL'Y'SIS CERTIFICATE OF

For: Shell Developement 140 St George's Terrace Perth W.A. 6000

Our ref: 1000.01.27272

Your ref: Date: 6.05.83

Description of Samples: Five water samples were received on the 20.04.83 for chemical analysis.

Method of Analysis: Chemical Data		Sample No.	
		No27	
рН		7.98	
Conductivity	(u siemens/cm)	115000	
T.F.R. (calcu	(lated)	73600	
		m g z'   1	m ∈quiv/l
Sodium	Na+	22200	965.6
Potassium	K+	410	10.49
Calcium	Ca++	520	25.95
Magnesium	Mg++	1300	107
Soluble Iron	Fe	1	_
Chloride	C 1-	37490	1056
Carbonate	003	<0.3	
Bi-Carbonate	HC03-	380.6	6.24
Sulphate	S04	2940	61.25
Nitrate	N03-	0.4046	0.006

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

### ANALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

### CERTIFICATE OF ANALYSIS

For: Shell Development 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 17.05.83

Bescription of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

Method of Analysis:		
a	B	
	8,87	
(), siemers/ $cm$ )		
	24400	
	m⊴∕ 1	m equiv/l
Na+	8100	352.3
K+	365	9.335
Ca++	53	2.645
Mg++	35	2.88
Fe	0.6	-
C 1-	9088	256
C03	267	8.9
HC03-	5893	96.6
804	475	9.896
N03-	<0.05	· · · · · · · · · · · · · · · · · · ·
	a 	a B (u siemens/cm) 38130 ulated) 24400 mg/l Na+ 8100 K+ 365 Ca++ 53 Mg++ 35 Fe 0.6 C1- 9088 C03 267 HC03- 5893 S04 475

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

TABLE 11Q

ANALABS

ANALYTICAL CHEMISTS

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

### CERTIFICATE OF ANALYSIS

For: Shell Development 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 17.05.83

m equiv/l

Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

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Method of Analysis:

Sample No.

Chemical Data

рН	8.39
Conductivity(u siemens/cm)	41750
T.F.R. (calculated)	26720
	mg/1

Sodium	Na+	8350	363.2
Potassium	K+	325	8.312
Calcium	Ca++	100	4.99
Magnesium	Mg++	70	5.76
Soluble Iron	ı Fe	0.9	_
Chloride	с 1-	11720	330
Carbonate	C03	12	0.4
Bi-Carbonate	HC03-	1964	32.2
Sulphate	804	475	9.896
Nitrate	N03-	<0.05	

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

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ANALABS

ANALYTICAL CHEMISTS

TABLE 11P

52 Murray Road Welshpool W.A. 6106 Tel: 458 7999

### CERTIFICATE OF ANALYSIS

For: Shell Development 140 St George's Terrace Perth W.A. 6000 Our ref: 1000.01.27272

Your ref: Date: 17.05.83

Description of Samples: Thirteen water samples were received on the 9.05.83 for chemical analysis.

Method of Analysis: Chemical Data		Sample No.	
		F	
рH		8.47	
Conductivity	(u siemens/cm)	40380	
T.F.R. (calcu	(lated)	25840	
		m⊗≓ 1	m equiv∕l
Sodium	Na+	8100	352.3
Potassium	K+	345	8.824
Calcium	Ca++	100	4.99
Magnesium	Mg++	75	6.172
Soluble Iron	Fe	0.15	-
Chloride	С 1-	11720	330
Carbonate	C03	18	0.6
Bi-Carbonate		1726	28.3
Sulphate	\$04	240	5
Nitrate	N03-	4.429	0.0714

Analyst: M.A. CHAPMAN A.P.T.C., A.R.A.C.I., A.A.I.M.M.

Analytical Chemist

## A P P E N D I X I

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### 1. PREPARATION OF SEDIMENT SAMPLES FOR EXTRACTION

The samples provided for geochemical studies are firstly, where necessary, carefully air dried. Then the samples are crushed to 1/8" chips using a van Gelder jaw crusher, and finally they are crushed to 0.1mm using an NV Tema grinder.

### 2. EXTRACTION OF SEDIMENT SAMPLES

Crushed sediment (maximum of 250g) and 320 mls of purified dichloromethane: methanol (10:1) were placed in a 500 ml conical flask. A double surface condenser was fitted to the flask, and the sample was then extracted under the influence of ultra-sonic vibration  $(60-70^{\circ}C)$  using a Buehler Ultramet II sonic bath for 2 hours. The solvent was then separated from the sediment using a large Buchner filtration system. The extract was recovered by careful evaporation of the solvent on a steam bath and weighed. The weight of extract was used to calculate %SOM(UNC) using the following formula:

$$%$$
SOM(UNC) = Wt. extract  $x \frac{100}{1}$ 

### 3. SEPARATION OF PETROLEUM INTO CONSTITUENT FRACTIONS

The extracts were separated into saturated, aromatic and NSO (asphaltenes plus resins) fractions by column chromatography on silicic acid. The crude extract was applied to the top of a silicic acid column (sample to adsorbent ratio 1:50) and the saturated compounds were eluted with <u>n</u>-pentane, aromatic compounds with a 50:50 mixture of ether and <u>n</u>-pentane, and finally the NSO fraction was eluted with a 20:1 mixture of methanol and dichloromethane. The neat fractions were recovered by careful removal of the solvent by fractional distillation and weighed.

The sum weight of the three fractions was used to calculate the %SOM using the following formula:

$$%SOM = \frac{Wt. AROM. + Wt. SAT. + Wt. NSO}{Wt. SEDIMENT EXTRACTED} \times \frac{100}{1}$$

This parameter can be used to assess the suitability of the sediments as source rocks according to the classification shown (later in this section) in the table "Classification of Source Rock Richness". The weight of saturated compounds was used to calculate the percentage of saturated compounds in the sediment according to the following formula:

$$%SaOM = \frac{Wt. Saturates}{Wt. Sediment Extracted} \times \frac{100}{1}$$

This parameter can be used to assess the suitability of the sediments as oil source rocks according to the classification shown in the table "Classification of Source Rock Richness".

The weight of each fraction was used to calculate the % by weight of each fraction in the extract according to the following formula:

% Fraction = 
$$\frac{\text{Wt. Fraction}}{\text{Wt. All Fractions}} \times \frac{100}{1}$$

The composition of the extracts can provide information about their levels of maturity and/or source type (LeTran et al., 1974; Philippi, 1974). Generally, marine extracts have relatively low concentrations of saturated and NSO compounds at low levels of maturity, but these concentrations increase with increased maturation. Terrestrially derived organic matter usually has a low level of saturates and large amount of aromatic and NSO compounds irrespective of the level of maturity.

### 4. GLC ANALYSIS OF SATURATED COMPOUNDS

Capillary GLC traces were recorded for each saturate fraction. The following information was obtained from these traces:

- (a) <u>n</u>-Alkane Distribution The C<sub>12</sub>-C<sub>31</sub> <u>n</u>-alkane distribution was determined from the area under peaks representing each of these <u>n</u>-alkanes. This distribution can yield information about both the level of maturity and the source type (LeTran et al., 1974).
- (b) Carbon Preference Index Two values were determined:

$$CPI(1) = \frac{(c_{23} + c_{25} + c_{27} + c_{29})Wt\% + (c_{25} + c_{27} + c_{29} + c_{31})Wt\%}{2 \times (c_{24} + c_{26} + c_{28} + c_{30})Wt\%}$$

$$CPI(2) = \frac{(c_{23} + c_{25} + c_{27})Wt\% + (c_{25} + c_{27} + c_{29})Wt\%}{2 \times (c_{24} + c_{26} + c_{28})Wt\%}$$

The CPI is believed to be a function of both the level of maturity (Cooper and Bray, 1963; Scalan and Smith, 1970) and the source type (Tissot and Welte, 1978). Marine extracts tend to have values close to 1 irrespective of maturity whereas values for terrestrial extracts decrease with maturity from values as high as 20 but don't usually reach a value of 1.

- (c) C<sub>21</sub>+C<sub>22</sub>/C<sub>28</sub>+C<sub>29</sub> This parameter provides information about the source of the organic matter (Philippi, 1974). Generally, a terrestrial source gives values <1.2 whereas a marine source results in values >1.5.
- (d) Pristane/Phytane Ratio This value was determined from the areas of peaks representing these compounds. The ratio renders information about the depositional environment according to the following scale (Powell and McKirdy, 1975):

<3.0 Marine depositional environment (i.e. reducing environment)</p>
3.0-4.5 Mixed depositional environment (i.e. reducing/oxidising environment)
>4.5 Terrestrial depositional environment (i.e. oxidising environment)

(e) Pristane/<u>n</u>-C<sub>17</sub> Ratio - This ratio was determined from the areas of peaks representing these compounds. The value can provide information about both the source type and the level of maturation (Lijmbach, 1975). Very immature crude oil has a pristane/<u>n</u>-C<sub>17</sub> ratio >1.0, irrespective of the source type. However, the following classification can be applied to mature crude oil:

<0.5	Marine source
0.5-1.0	Mixed source
>1.0	Terrestrial source

In the case of sediment extracts these values are significantly higher and the following classification is used:

<1.0	Marine source
1.0-1.5	Mixed source
>1.5	Terrestrial source

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- (f) Phytane/<u>n</u>-C<sub>18</sub> Ratio This ratio was determined from the areas of peaks representing these compounds. The value usually only provides information about the level of maturity of petroleum. The value decreases with increased maturation.
- (g) Relative Amounts of <u>n</u>-Alkanes and Naphthenes Since <u>n</u>-alkanes and naphthenes are the two dominant classes of compounds in the saturate fraction, a semi-quantitative estimate of the relative amounts of these compounds was made. This information can be used to assess the degree of maturation and/or the source type of the petroleum (Philippi, 1974; Tissot and Welte, 1978). Very immature petroleum has only small proportions of <u>n</u>-alkanes, but as maturity increases the relative amount of <u>n</u>-alkanes increases. In addition, terrestrial petroleum has a greater proportion of high molecular weight naphthenes than marine petroleum.

#### 5. TOC DETERMINATIONS

The total organic carbon value (TOC) was determined on the unextracted sediment sample. The value was determined by treating a known weight of sediment with dilute HCl to remove carbonate minerals, and then heating the residue to 1700°C (Leco Induction Furnace) in a atmosphere of pure oxygen. The carbon dioxide produced was adsorbed on a "Carbosorb" tower. The weight of carbon dioxide produced was then used to calculate %TOC in the sediment.

#### 6. SOLUBLE/TOTAL ORGANIC CARBON RATIOS

The ratios of SOM(mg)/TOC(g) and SAT(mg)/TOC(g) were determined from the appropriate data. The SOM(mg)/TOC(g) ratio can be used as a maturation indicator, especially if the parameter is plotted against depth for a given sedimentary sequence. In an absolute sense it is less reliable as a maturation indicator, although previous work (Tissot et. al., 1971; LeTran et. al., 1974) suggest that the following criteria can be used to determine maturity with this parameter:

<50 Low maturity
50-100 Moderate maturity
>100 High maturity

The ratios of SOM(mg)/TOC(g) and SAT(mg)/TOC(g) can be used collectively to provide information about source type. For example, if SOM(mg)/ TOC(g) is >100, suggesting a high level of maturity, but the SAT(mg)/ TOC(g) <20 it is very likely that the organic matter is terrestrial. Conversely, the same SOM(mg)/TOC(g) value with a SAT(mg)/TOC(g) value >40 suggests a marine source type.

#### 7. ROCK-EVAL PYROLYSIS

Rock-Eval pyrolysis is carried out by placing approximately 100mg of the crushed sample into a crucible and then subjecting it to the following pyrolysis cycle:

- Stage (iii)- Sample heated from 300°C to 550°C at 25°C/minute to produce petroleum from kerogen (S<sub>2</sub> peak). The furnace is maintained at 550°C for one minute. Carbon dioxide produced during this pyrolysis up to 390°C (550°C in the case of the carbonate-free sediment) is absorbed on a special column;
- Stage (iv) During cool-down period the carbon dioxide produced during pyrolysis is measured (S3 peak).

The units used for Rock-Eval data are as follows:

 $S_1$ ,  $S_2$ ,  $S_3$  = kg/tonne of rock  $T_{max}$  =  $^{O}C$ Hydrogen Index = mg HC/g TOC Oxygen Index = mg CO<sub>2</sub>/g TOC

Rock-Eval data is most commonly used in the following manner:

- (i) S<sub>1</sub> indicates the level of oil and/or gas already generated by the sample.
- (ii) S1+S2 referred to as the genetic potential this parameter is used for source rock evaluation according to the following criteria:

<2	kg/tonne	Poor
2-6	kg/tonne	Moderate
>6	kg/tonne	Good

- (iii)  $S_1/(S_1+S_2)$  this parameter is the production index which is a measure of the level of maturity of the sample.
- (iv) T the temperature corresponding to the S<sub>2</sub> maxima. This temperature increases with increasingly mature sediments.
- (v) HI, OI the hydrogen ([S<sub>2</sub>x100]/TOC) and oxygen ([S<sub>3</sub>x100]/TOC) indices when plotted against one another provide information about the type of kerogen contained in the sample and the maturity of the sample.

### 8. WHOLE OIL GAS CHROMATOGRAPHIC ANALYSIS

This analysis was carried out under the following conditions: Instrument = Varain Aerograph 2740; coloumn =  $25m \times 0.2mm$  I.D. WCOT capillary column with SP2100 stationary phase; temperature program = 2 mins at  $-20^{\circ}$ C then programmed to  $280^{\circ}$ C at  $4^{\circ}$ C/min; detector temperature =  $310^{\circ}$ C; injector temperature =  $300^{\circ}$ C; injection mode 80:1 split; carrier gas = hydrogen at 35 cm/sec.

#### REFERENCES

- Alexander, R., Kagi, R.I. and Woodhouse, G.W. "Measurement of thermal maturation of petroleum by proton magnetic resonance spectroscopy". Nature, <u>276</u>, 1978, 598.
- Alexander, R., Kagi, R.I. and Woodhouse, G.W. "A new method for measuring the maturity of petroleum in source rocks". APEA J., <u>19</u>, 1979, 90-93.
- Cooper, J.E. and Bray, E.E. "Apostulated role of fatty acids in petroleum formation". Geochim. Cosmochim. Acta, 27, 1963, 1113-1127.
- Gransch, J.A. and Eisma, E. "Characterization of the insoluble organic matter of sediments by pyrolysis". <u>Advances in Organic Geochemistry</u>, 1966, 407-426.
- Hunt, J.M. "Geochemistry of petroleum". Am. Assoc. Pet. Geol. Continuing Education Lecture Series.
- Lijmbach, G.W.M. "On the origin of petroleum". Proc. 9th World Petroleum Congress, 2, 1975, 357-369.
- LeTran, K., Connan, J. and Van der Weide, B. "Diagenesis of organic matter and occurrence of hydrocarbons and hydrogen sulphide in the S.W. Aquitaine Basin". Bull. Centre Rech., Pau-SNPA, 8, 1974, 111.
- Philippi, G.T. "The influence of marine and terrestrial source material on the composition of petroleum". Geochim. Cosmochim. Acta, <u>38</u>, 1974, 947.
- Powell, T.G. and McKirdy, D.M. "Geological factors controlling crude oil composition in Australia and Papua New Guinea". Amer. Assoc. Petrol. Geol. 59, 1975, 1176.
- Scalan, R.S. and Smith, J.E. "An improved measure of the odd-even predominance in the normal alkanes of sediment extracts and petroleum". Geochim. Cosmochim. Acta, 34, 1970, 611-620.
- Stahl, W.J. "Carbon and nitrogen isotopes in hydrocarbon research and exploration". Chem. Geol., <u>20</u>, 1977, 121-149.
- Stahl, W.J. "Source rock-crude oil correlation by isotopic type-curves". Geochim. Cosmochim. Acta, <u>42</u>, 1978, 1573-1577.

- Tissot, B. et al. "Origin and evolution of hydrocarbons in early Toarcian shales, Paris Basin, France". Amer. Assoc. Petrol. Geol., <u>55</u>, 1971, 2177.
- Tissot, B. et al. "Influence of nature and diagenesis of organic matter in the formation of petroleum". Amer. Assoc. Petrol. Geol., <u>58</u> 1974, 499.
- Tissot, B. and Welte, D.H. "Petroleum Formation and Occurrence". Springer-Verlag. Berlin Heidelberg New York, 1978.
- Welte, D.H., et al., "Correlation between petroleum and source rock". Proc. 9th World Petroleum Congress, <u>2</u>, 1975, 179-191.

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FIGURE 1 APRIL, 198	33			НҮ	/ DROCARBO		— L C се воск		ION				SHELL VOLADOR #1
LIME DOLOMITE CLAY SILT SAND COAL IGNEOUS META EVAPORITE LITHOLOGY		CUTTIN CONV. SWC SAMPLE POOR FAIR GOOD	CORE	S <sub>1</sub> = VC S <sub>2</sub> = HC S <sub>1</sub> + S <sub>2</sub> S <sub>3</sub> = OF	DLATILE HYDR	OCARBONS POTENTIA PROCARBON	(HC) × AL PI = P · I PI = $\frac{S}{S_1}$	POPULICITION	HI = HYDROGEN INDEX HI = $\frac{S_2}{TOC} \times 100$ OI = OXYGEN INDEX OI = $\frac{S_3}{TOC} \times 100$	)	P O W R U T O W I A VITRI	R M A L C C 1	INDIGENOUS KEROGEN ALGINITE EXINITE VITRINITE INERTINITE INDIGENOUS
(Percent) 20 40 80 80	Depth	(Percent	t of Rock)	10 8	S1; S1+S2	180 1	S3 2 3 4	PI .2.4	■ HI ▲ OI 100 200 300	Tmax <sup>O</sup> C 440 460		NITE NCE 1,8	KEROGEN S 25 50 75 IND
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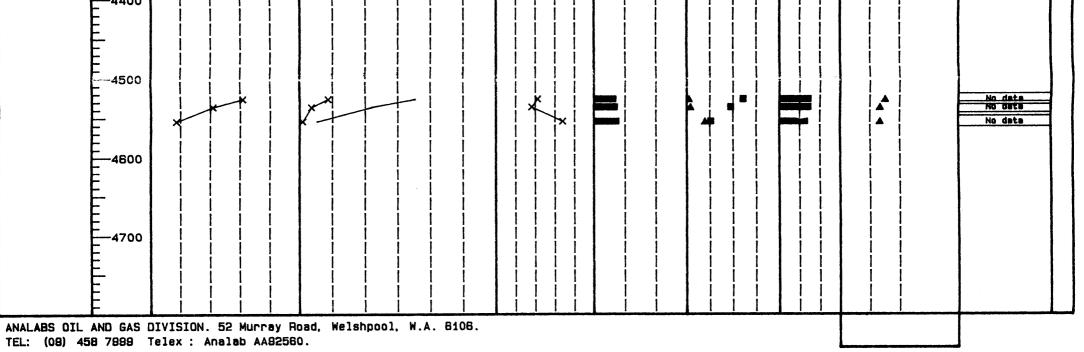


FIGURE 2 APRIL, 198	83					 НҮD	ROCA		N 1 so						_UA	TION							SHELL VOLADOR #1	
LIME DOLOMITE CLAY SILT SAND COAL IGNEOUS META ZZ EVAPORITE	■ IN METRES	CUT CON SAMP POOP FAIF	V. COI LE Q A A D	RE UAL	ITY			EXT TOTAL SATUR AROMA NITRO ASPHA	EXTR ATE TIC GEN-S	AACT (SAT (ARO SULP E (A	(E. ABLE .) H M.) HUR- SPH)	. 0 . : ORI IYDRI H.C	M.) gani dcar gen	C MA BON (N.S	(H.(	2.) T( H	DTAL .C. ON .C.		PARE OHL WHZDOW	OHL WHZOOW	WEF GASYCOND	Ē	INDIGENO KEROGEN ALGINITE EXINITE VITRINITE INERTINITE	N I
LITHOLOGY (Percent) 20 40 60 80	Depth	(Perc	NIC C ent o 5 10	f Roc	sk)	E.O.M	(ppm)		000	<u>Sat</u> Arom	■ ▲ 1 - 2	ASPH NSO	Nor	H.C. H.	c.	TOC -	<b>5</b> 0	<b>EOM</b> 100			RINI		INDIGENOUS KEROGEN 25 50 75	S X IND
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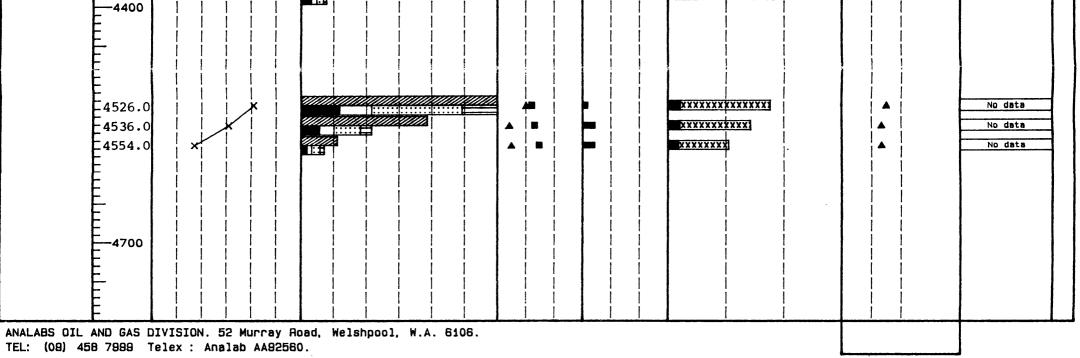
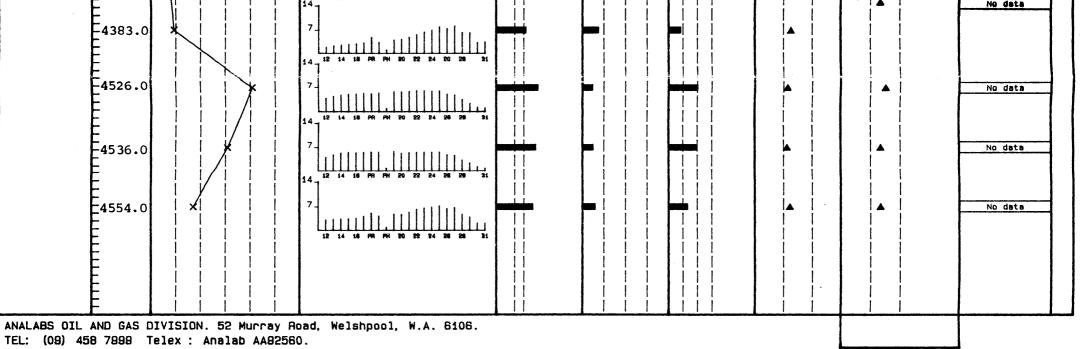


FIGURE 3			ANA-LOG	SHELL
APRIL, 1983	3		HYDROCARBON SOURCE ROCK EVALUATION	VOLADOR #1
CLAY SILT SAND COAL IGNEOUS META EVAPORITE	IN METRES IN FEET	SAMPLE TYPE CUTTINGS CONV. CORE SWC SAMPLE QUALITY POOR FAIR GOOD	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	T H H KEROGEN M A C C 1 C I I NERTINITE
LITHOLOGY (Percent) 20 40 60 80	Depth	ORGANIC CARBON (Percent of Rock) 2,5 5 10 20 40	ALKANE & DISTRIBUTION PRIST/PHYT PRIST/NC17 21+22/28+29 CPI REFLECTAN 2 4 6 8 10 12 2 4 6 1 2 3 4 5 1 1,5 .6 1,2	
	3549.9 3645.0 3673.5 3800 3900 4100			No data No data







**OIL AND GAS DIVISION** 

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